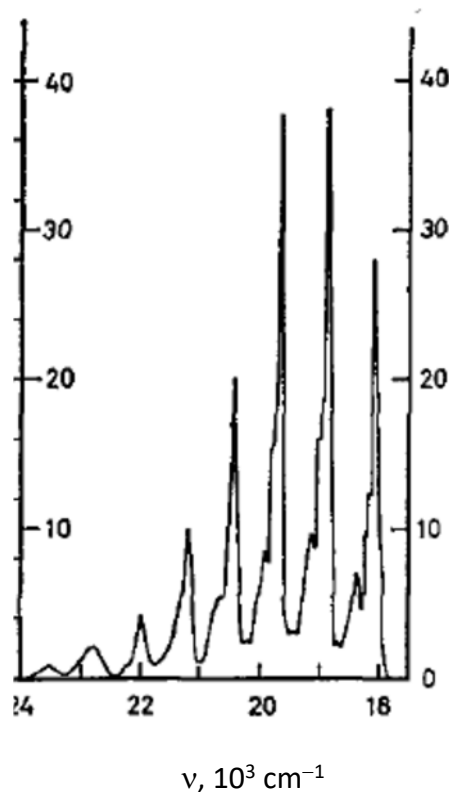


Ch153a
 Winter 2019
 Due 18 January, 2019

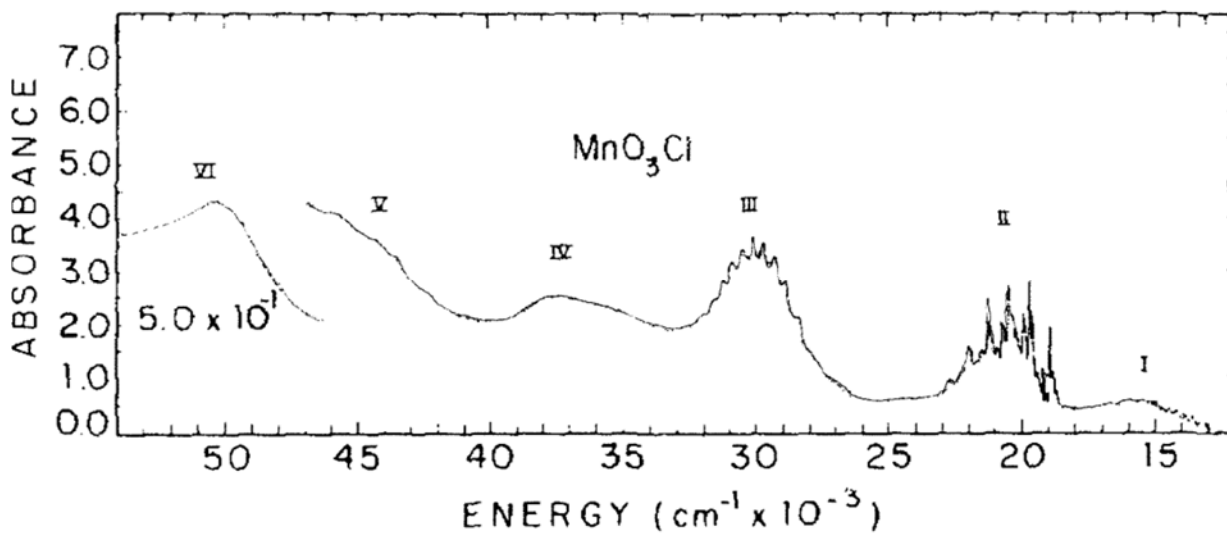
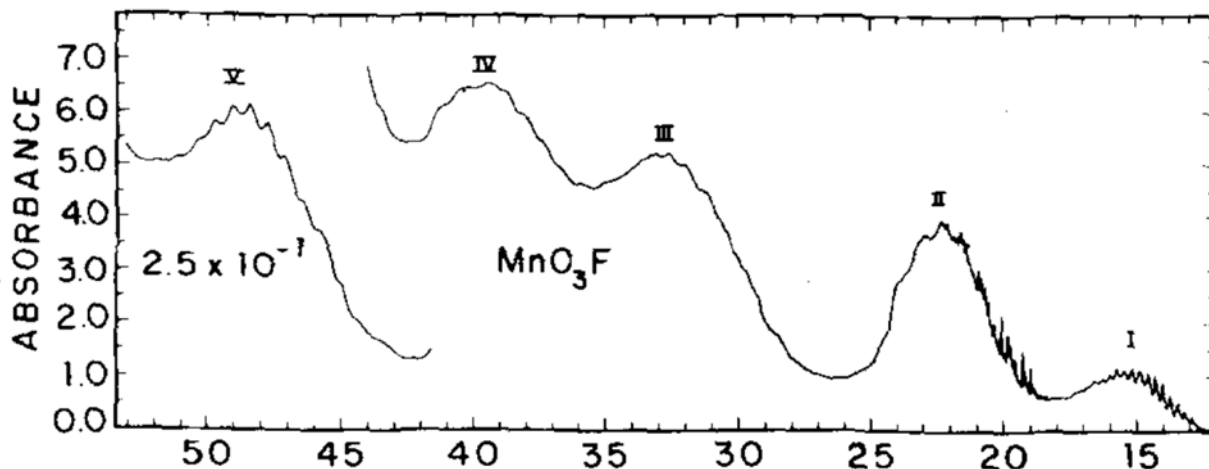
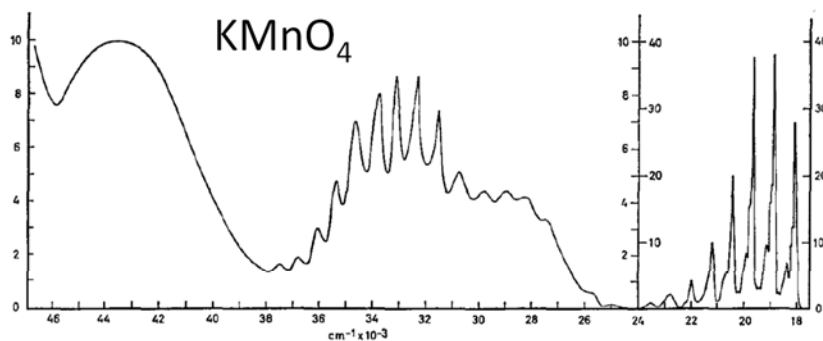
Problem Set 2

(1) The 4K single crystal absorption spectrum of KMnO_4 doped into a KClO_4 lattice and a table of peak positions are shown below.

Peak Position (cm^{-1})	Relative Intensity
18,072	74
18,842	100
19,602	96
20,380	59
21,145	27
21,915	10



- (a) Perform a Franck-Condon analysis of the vibrational fine structure in the absorption band.
- (b) What S_{HR} -value gives the best fit to the observed spectrum?
- (c) If the force constant for the distorting mode is 5.91 mdyne/\AA , give the magnitude of the distortion in the normal mode (ΔQ).
- (d) Use group theory to define the normal mode in terms of bond stretching coordinates to estimate the magnitude of the distortion in the individual bonds.
- (2) Using the basis functions for T_d coordination with C_3 quantization developed in Problem Set 1, draw a correlation diagram relating the d -orbital splitting in a tetrahedral MX_4 molecule to that in a MX_3Y molecule with C_{3v} symmetry (X and Y are strong σ and π donor ligands).
- (3) The 4K single crystal absorption spectrum of KMnO_4 doped into a KClO_4 lattice, and the gas-phase absorption spectra of MnO_3F and MnO_3Cl are shown on the following page.



The gas-phase vibrational spectra of MnO_3F and MnO_3Cl exhibit symmetric Mn-O stretches at 905 and 892 cm^{-1} , respectively, and symmetric Mn-X vibrations at 721 ($X = \text{F}$) and 460 ($X = \text{Cl}$) cm^{-1} . The vibrational fine structure in bands I-IV of MnO_3X is summarized in the following table.

Band	MnO ₃ F	MnO ₃ Cl
I	851 cm ⁻¹	825 cm ⁻¹
II	819 cm ⁻¹	783 cm ⁻¹
III	653 cm ⁻¹	402 cm ⁻¹
IV	655 cm ⁻¹	618 cm ⁻¹

The assignments of the MnO₄⁻ absorption bands were discussed in class. Propose assignments for bands I, II, III, and IV in MnO₃F and MnO₃Cl.

- (4) Consider a hypothetical trigonal (C_{3v}) Co-oxo complex with ancillary ligands L, Co(O)L₃ⁿ⁺.
- Develop a ligand field splitting diagram for complexes of this type with Co(III) and Co(V) oxidation states.
 - Predict the ground-state electronic structure (term symbol including spin) for each complex.
 - Identify the Co-oxo bond order for each complex.